



Valorisation of char residues from biomass gasification in adsorption applications

V. Benedetti, E. Cappelletti, I. Moresio, E. Zanin, M. Baratieri

Heraklion, 27.06.2019



Biomass gasification

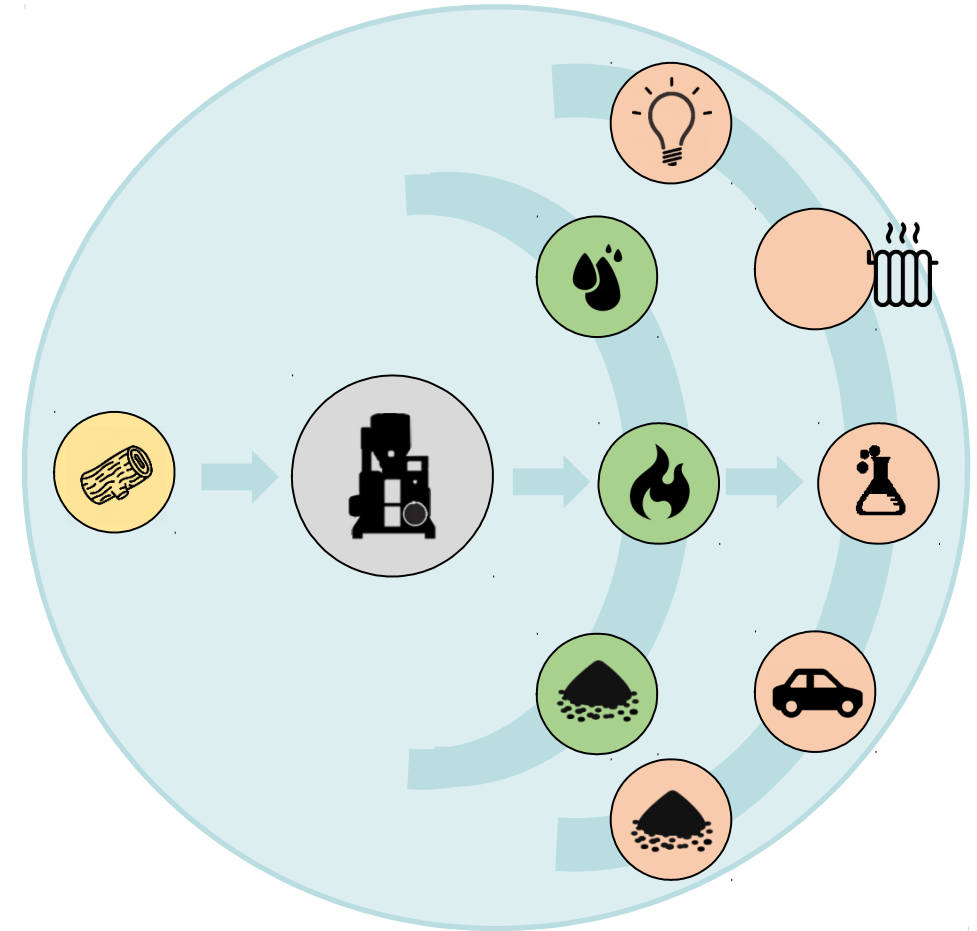
- Gas
- Tar (liquid)
- Char (solid)



South Tyrol: about **1300 tons/year** of char disposed of as industrial waste with a high cost for disposal (140 - 150 €/ton)

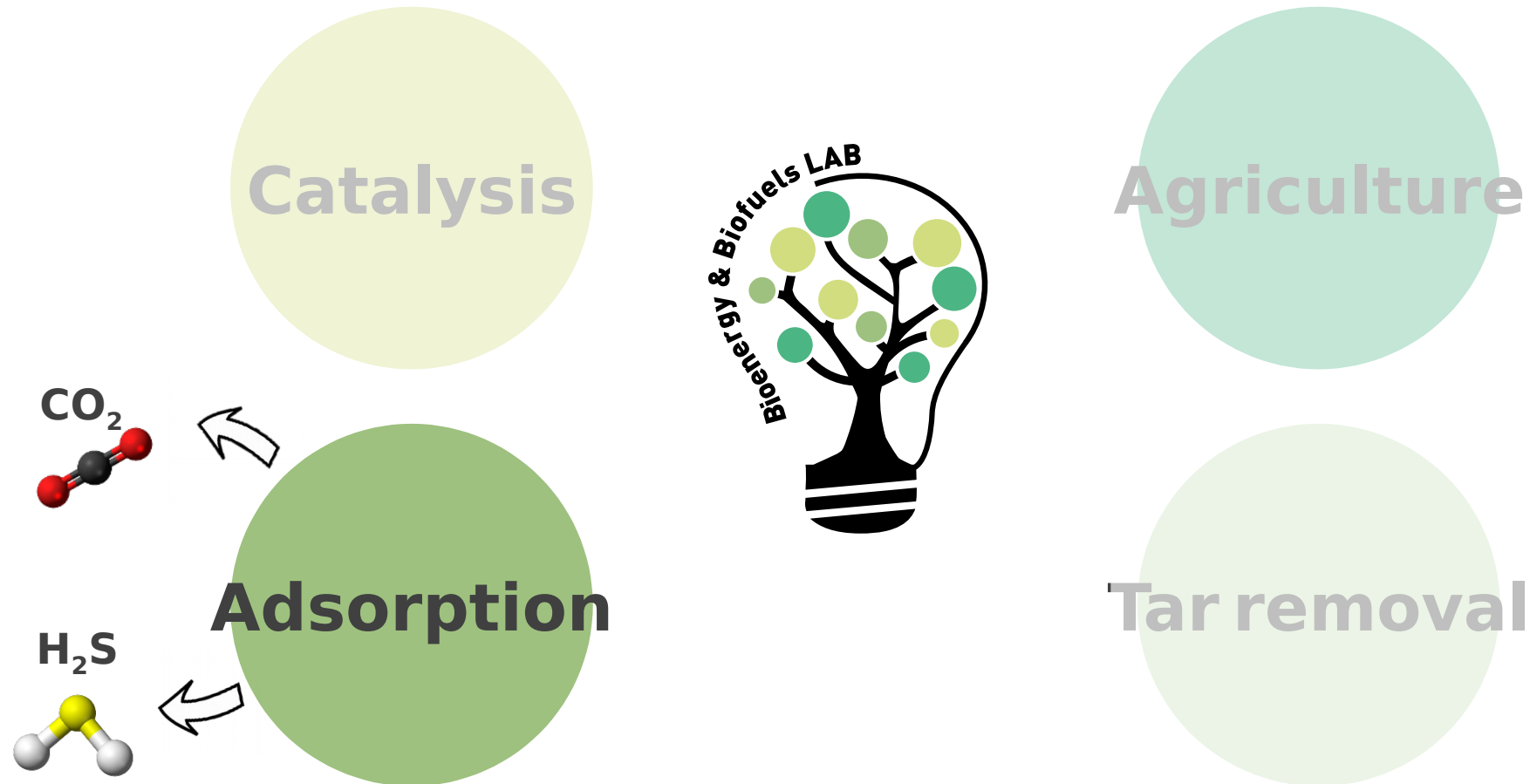


Valorization



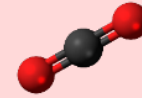


Char valorization at UNIBZ

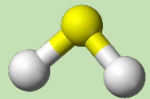




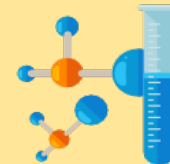
Char collection
and characterization



CO₂
adsorption



H₂S
adsorption



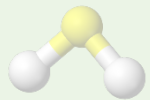
Other
applications



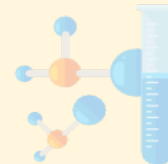
Char collection
and characterization



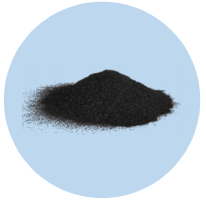
CO₂
adsorption



H₂S
adsorption



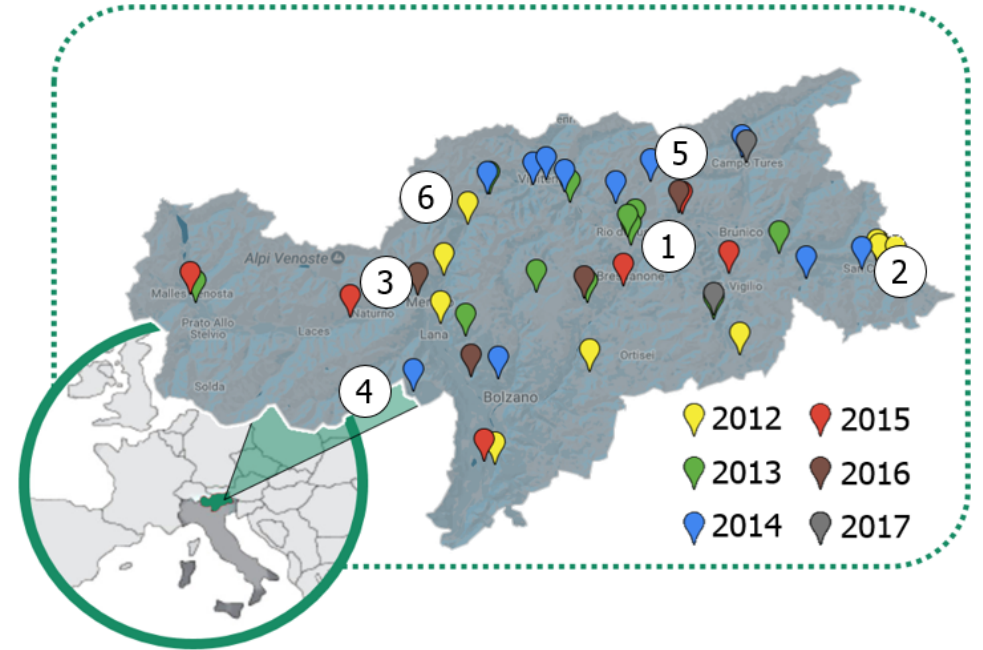
Other
applications



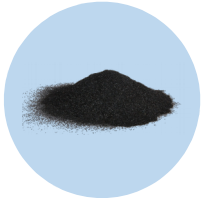
Gasification technologies

	Feedstock	Technology	Electric power kW_{el}	Thermal power kW_{th}	T $^{\circ}\text{C}$
1	Wood chips	Dual-stage	50	110	~900
2	Wood chips	Dual-stage	280	540	~850
3	Pellets	Rising co-current	180	270	~700
4	Wood chips	Downdraft	150	260	~650
5	Wood chips	Downdraft	296	550	~800
6	Wood chips	Downdraft	45	120	~650

- Scanning electron microscopy
- Small angle X-ray scattering
- Thermogravimetric analysis

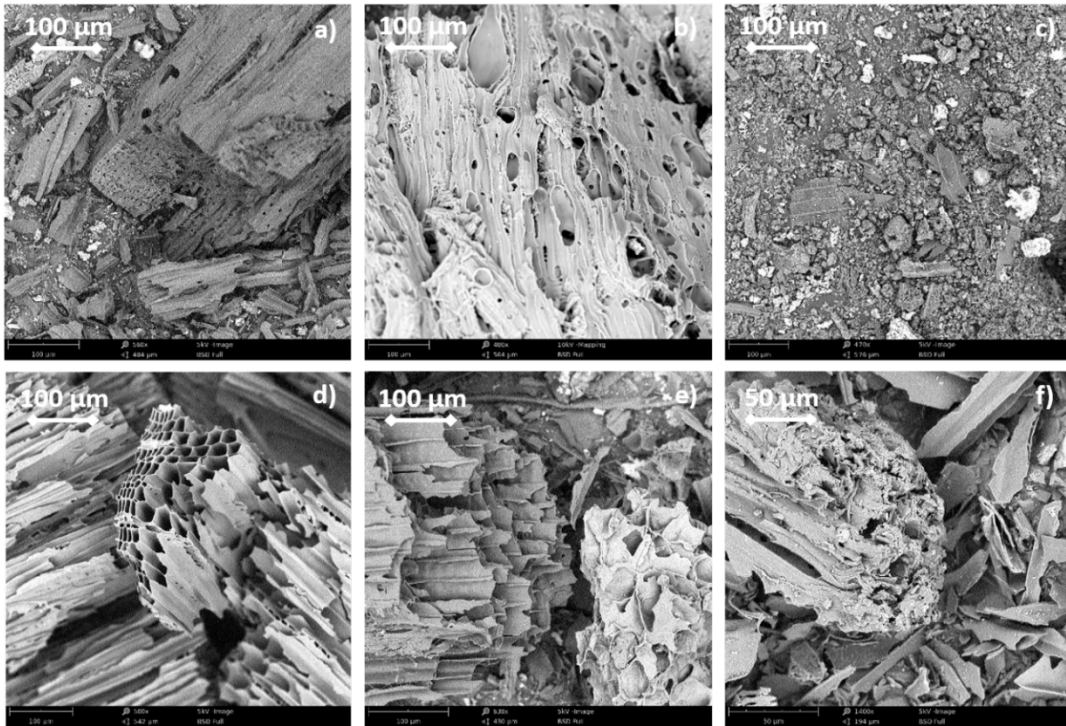


V. Benedetti et al., Characterization of char from biomass gasification and its similarities with activated carbon in adsorption applications, Appl. Energy, 227 (2018) 92-99.

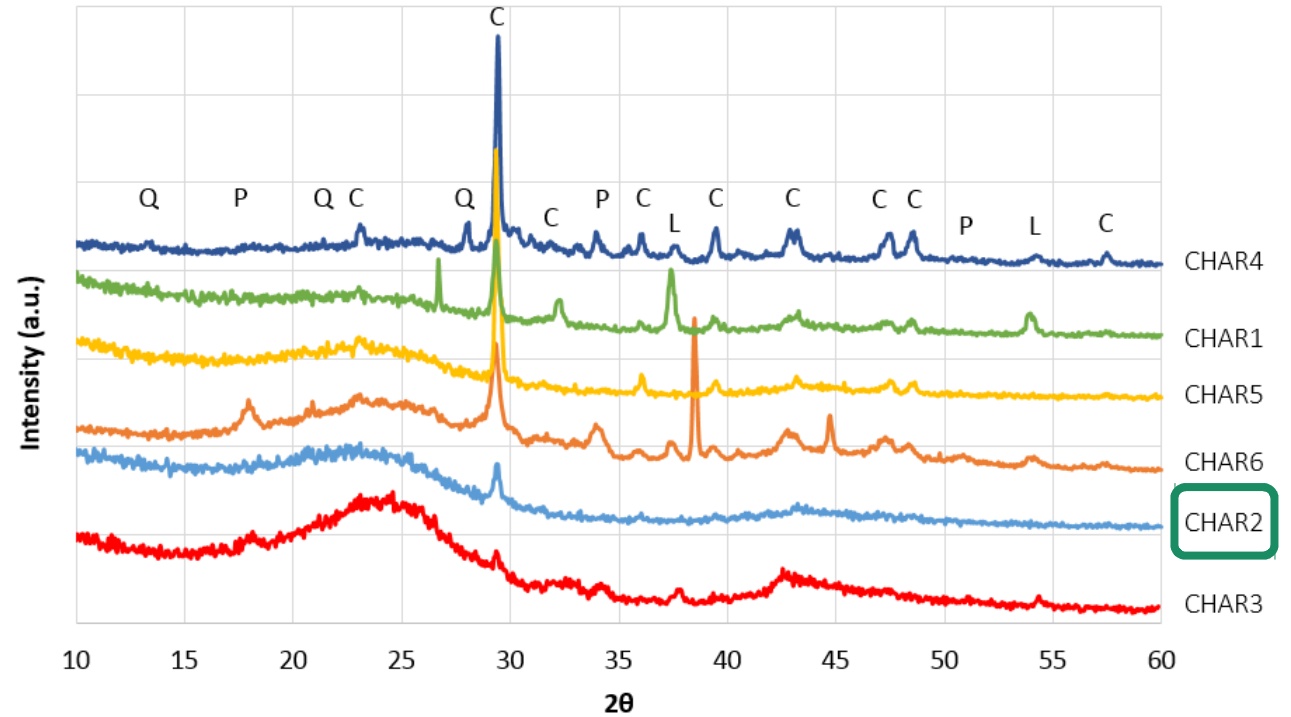


Differences among chars

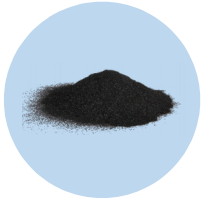
SEM



XRD



Q: quartz (SiO_2), P: portlandite (Ca(OH)_2), C: calcite (CaCO_3), L: lime (CaO)



Elemental analysis (% wt_{dry})

Sample	C %wt _d ry	H %wt _d ry	N %wt _d ry	S %wt _d ry	O %wt _d ry	Ash %wt _d ry
CHAR1	81.13	0.23	0.28	0.35	3.74	14.62
CHAR2	91.39	0.72	0.26	0.57	3.43	4.20
CHAR3	81.17	0.25	0.61	0.27	1.89	16.08
CHAR4	48.12	0.49	0.23	0.32	1.64	49.52
CHAR5	80.64	0.55	0.20	0.20	2.79	15.80
CHAR6	68.63	0.33	0.83	0.32	2.05	27.84

1 Highest surface area
 Low amount of ash

2 High surface area
 Lowest amount of ash

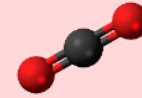
Surface area

	S _{BET} m ² / g	pore size nm	pore volume cm ³ /g
CHAR1	603	3.88	0.30
CHAR2	297	4.50	0.26
CHAR3	403	4.70	0.50
CHAR4	183	4.90	0.25
CHAR5	427	4.40	0.39
CHAR6	252	4.54	0.24

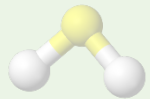
ADSORPTION
CATALYSIS



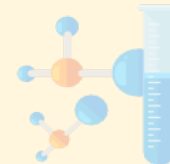
Char collection
and characterization



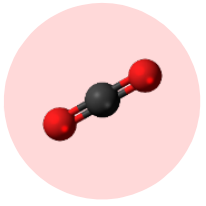
CO₂
adsorption



H₂S
adsorption

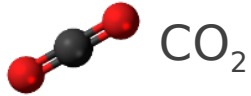


Other
applications



Materials and methods

Adsorptive:
e:
Adsorbent:



5 pure chars

2 activated chars

2 AC

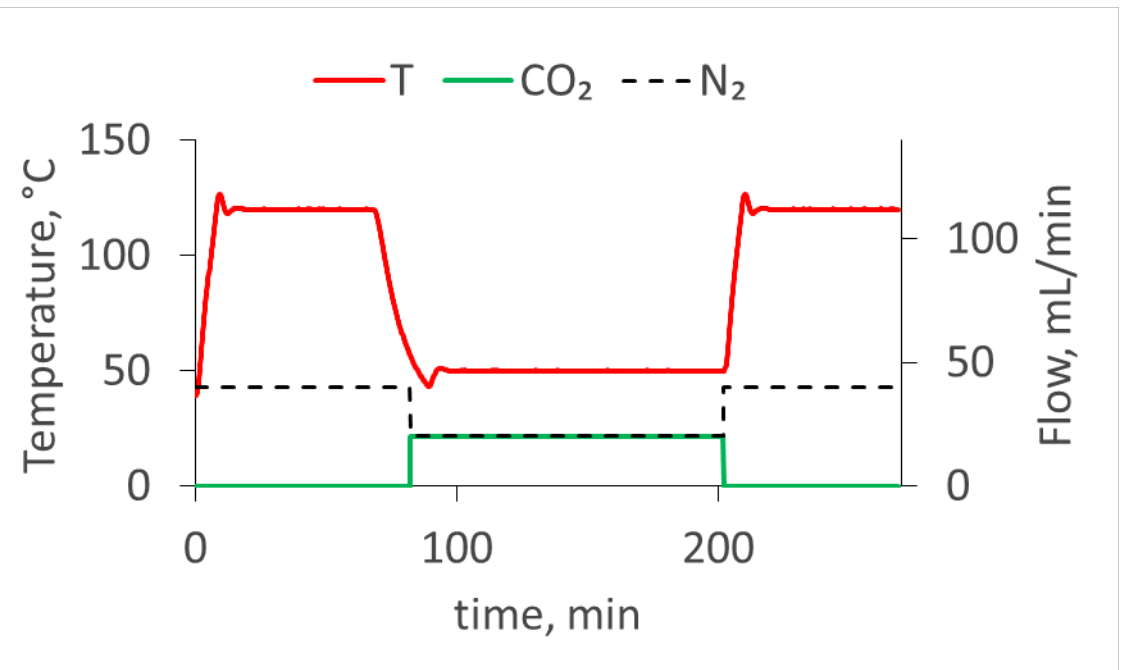


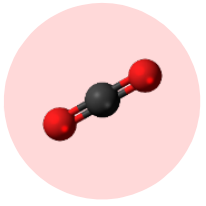
- KOH - ZnCl₂
- N₂
- 600 °C
- 1 hour



Thermo-gravimetric tests
run in a Jupiter
STA449-F3 (Netzsch)

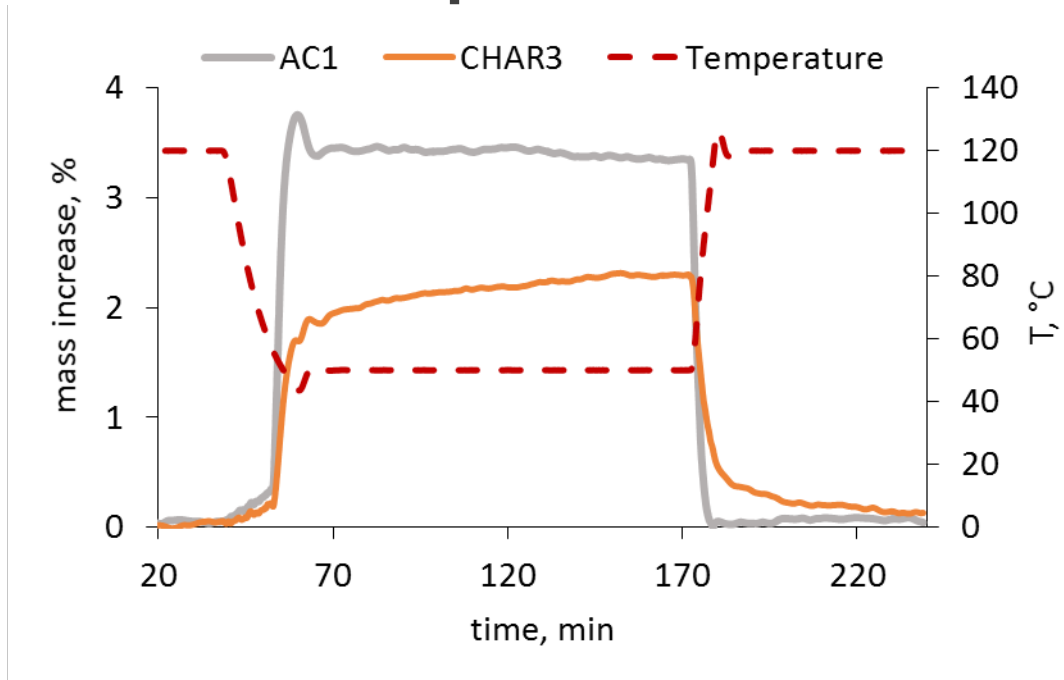
- T_{ads} = 50 - 75 - 100 °C
- CO₂:N₂ = 1:1 - 1:4





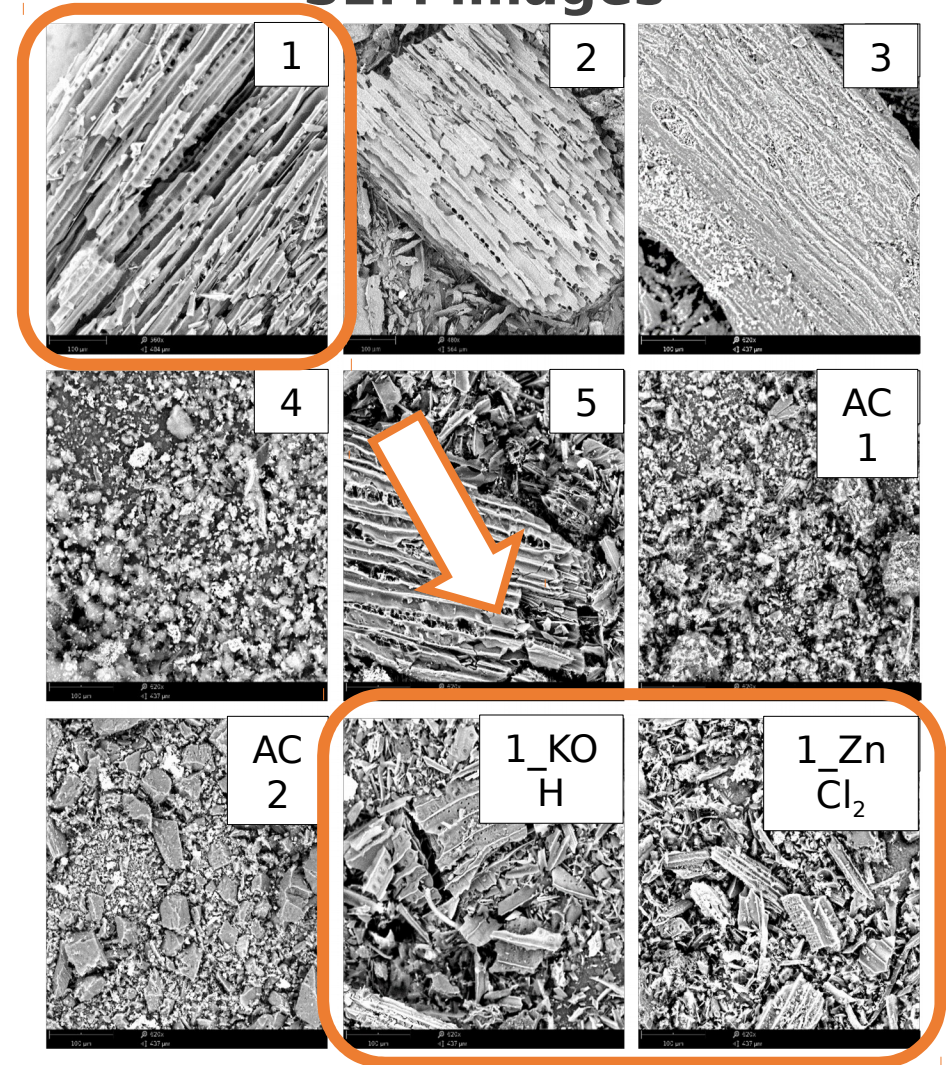
Results

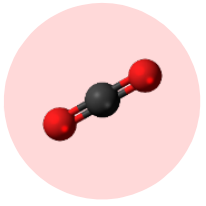
Adsorption curves



V. Benedetti et al., *CO₂ adsorption study on pure and chemically activated chars derived from commercial biomass gasifiers*, *J. CO₂ util.*, 33 (2019) 46-54.

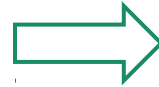
SEM images



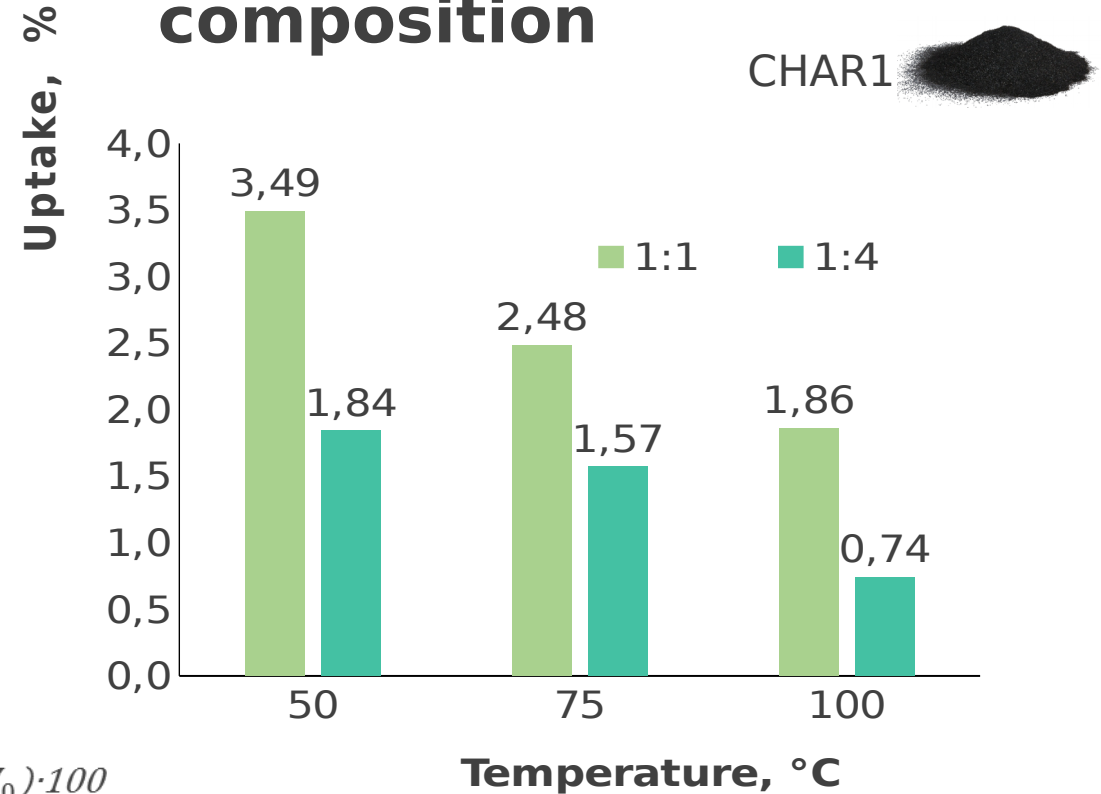


Results

	Sample	Uptake, %	N ₂ uptake, %
Pure char	CHAR1	3.49	0.20
	CHAR2	3.04	0.14
	CHAR3	2.09	0.08
	CHAR4	1.69	0.65
	CHAR5	2.75	0.17
Activated char	AC1	3.01	0.10
	AC2	2.13	0.07
	CHAR1_K	3.73	
	OH		0.35
	CHAR1_Z nCl2	3.03	0.13

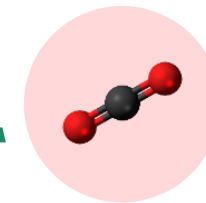


Effect of T and gas composition

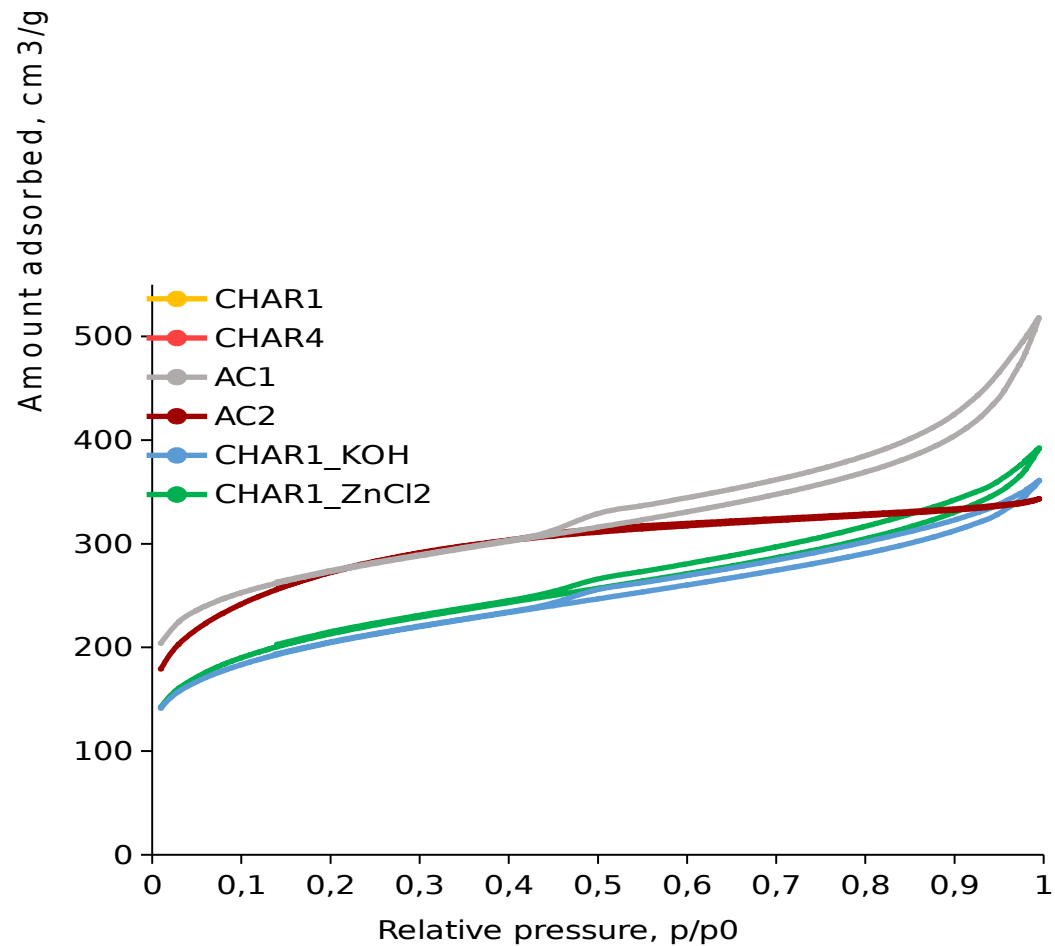


Literature 2.50 - 10.70

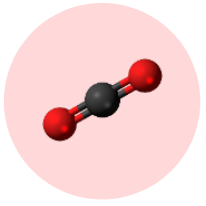
$$Uptake = \frac{(W_{end} - W_0) \cdot 100}{W_0}$$



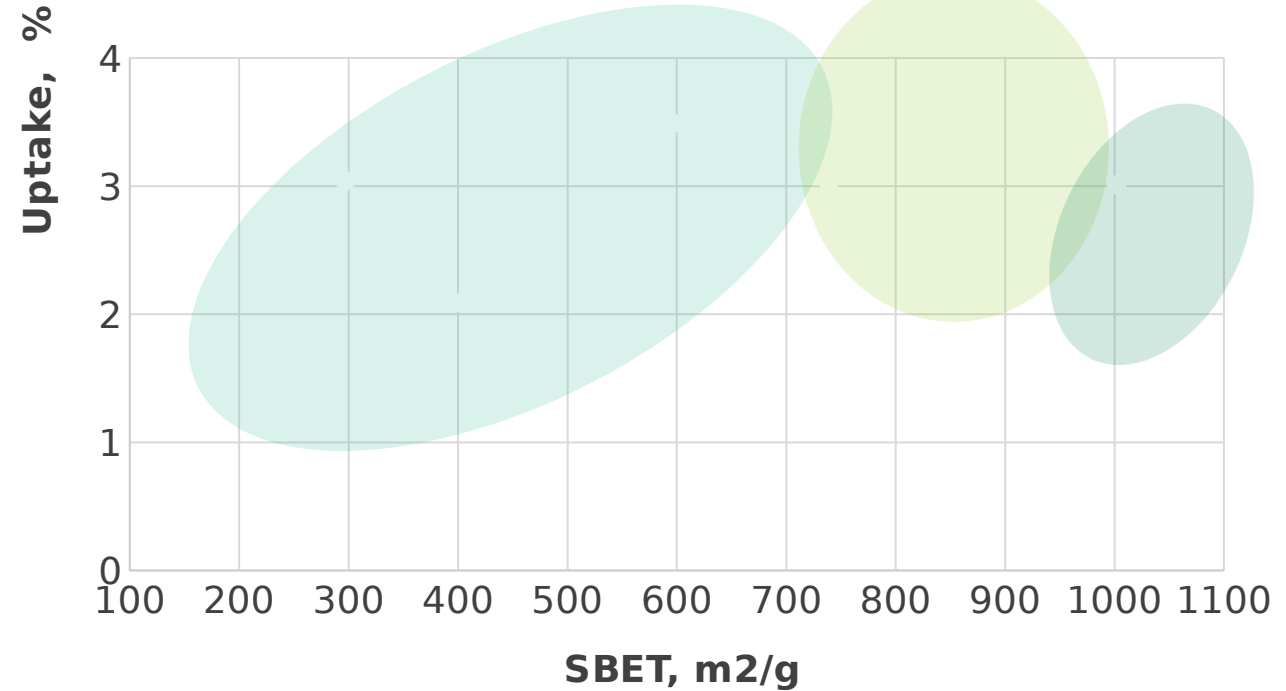
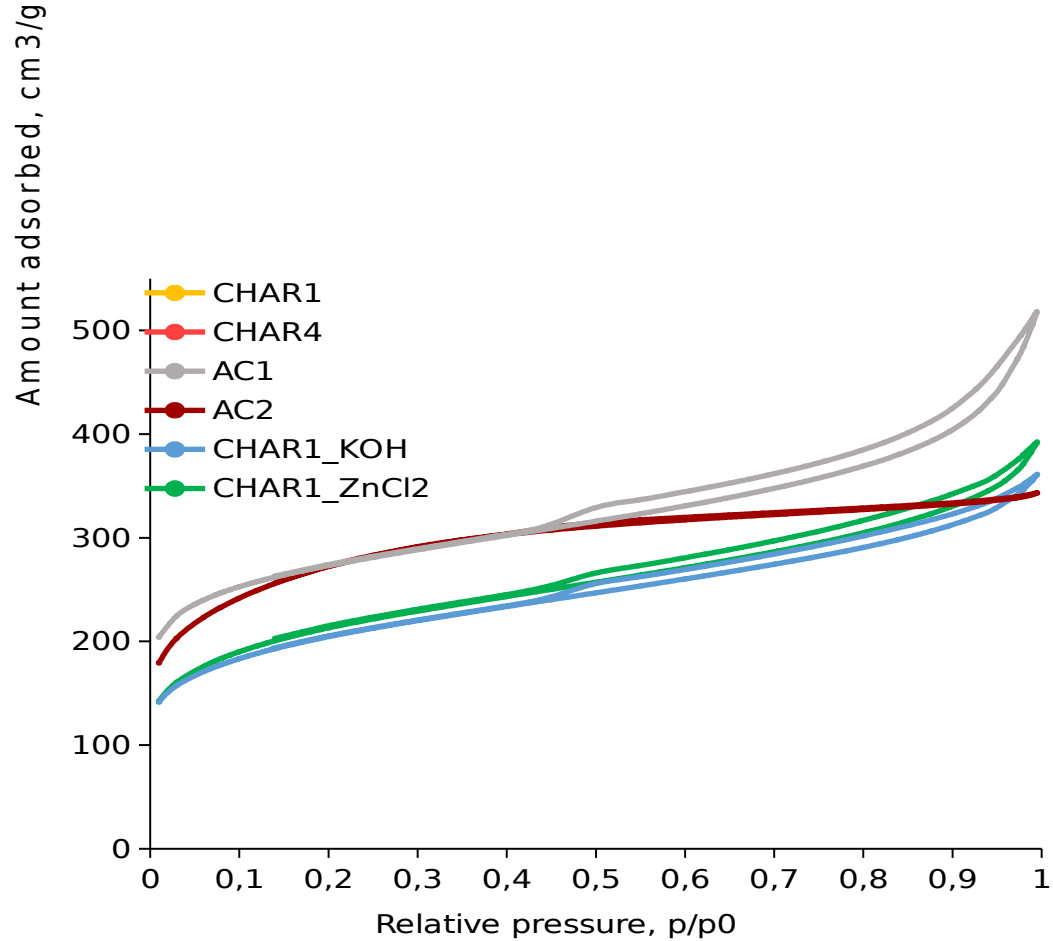
Results - physisorption

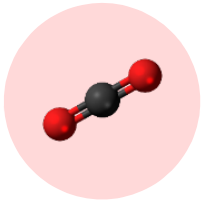


	S_{BET} m^2/g	Pore size nm	Pore volume cm^3/g	μ -pore volume cm^3/g
CHAR1	603	3.88	0.30	0.05
CHAR2	297	4.50	0.26	0.02
CHAR3	403	4.70	0.50	0.02
CHAR4	183	4.90	0.25	n.a.
CHAR5	427	4.40	0.39	0.06
AC1	1002	6.10	0.51	0.22
AC2	984	2.90	0.25	0.14
CHAR1_KOH	774	4.57	0.41	0.11
CHAR1_ZnCl2	739	4.85	0.37	0.11



Results - physisorption

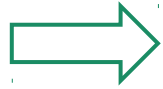




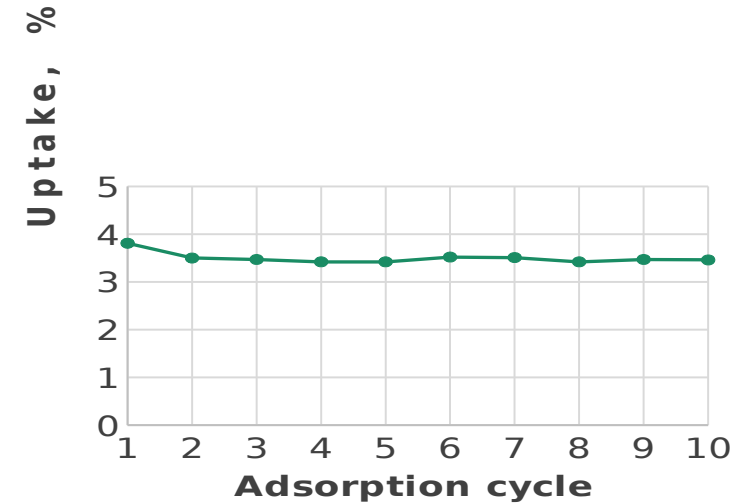
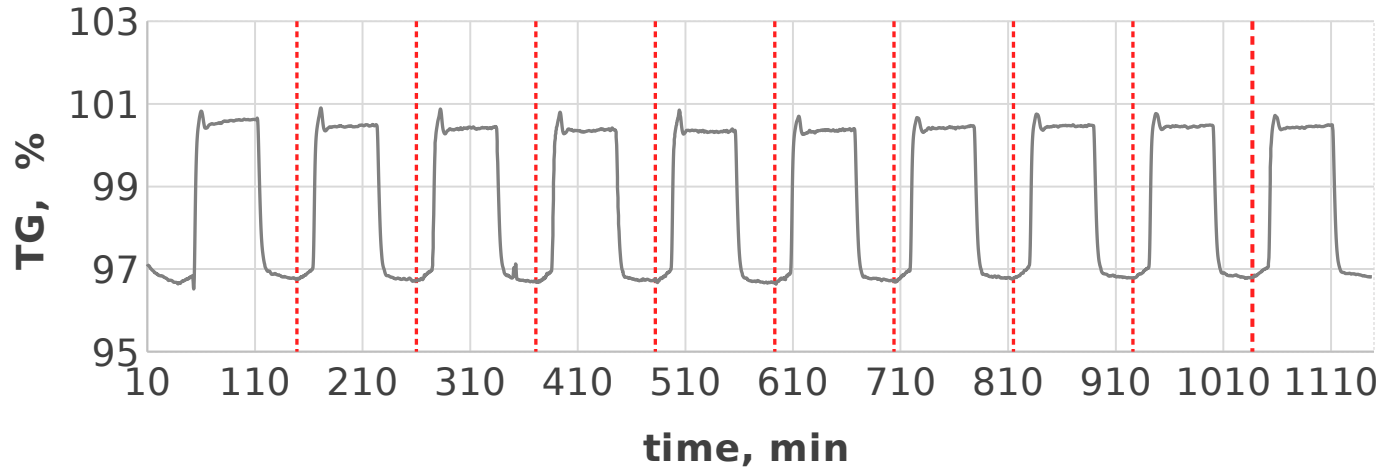
Results



CHAR1_KOH



Adsorption/desorption cycles to test loss of char adsorption capacity



- High selectivity
- High adsorption capacity
- Good adsorption/desorption kinetics
- Stable adsorption capacity after repeated cycles

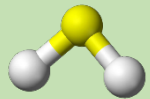
Effective CO₂ adsorbent



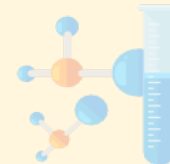
Char collection
and characterization



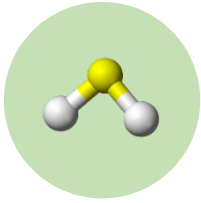
CO₂
adsorption



H₂S
adsorption

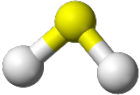




Other
applications



Materials and methods

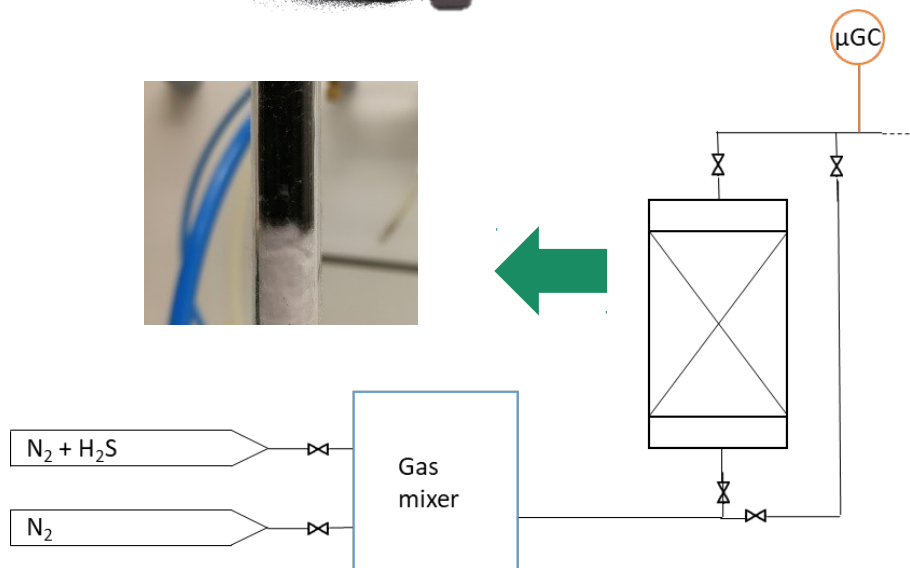
Adsorptiv
 e:
 Adsorben
 t:

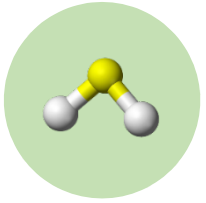
 H₂S
 5 pure chars
 2 AC

1 - Adsorption capacity

- Fixed bed reactor - quartz
- Char bed height: 2.5 cm (150 - 200 mg)
- H₂S + N₂: 100 NmL/min
- H₂S: 250 ppm
- T_{amb}
- Micro-GC for gas analysis at the

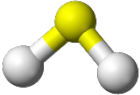

$$m_{ads} = f \frac{M P Q}{R T m_{char}} \int_0^{t_{fin}} (y_{in} - y_{out}) dt$$





Materials and methods

Adsorptiv
 e:
 Adsorben
 t:

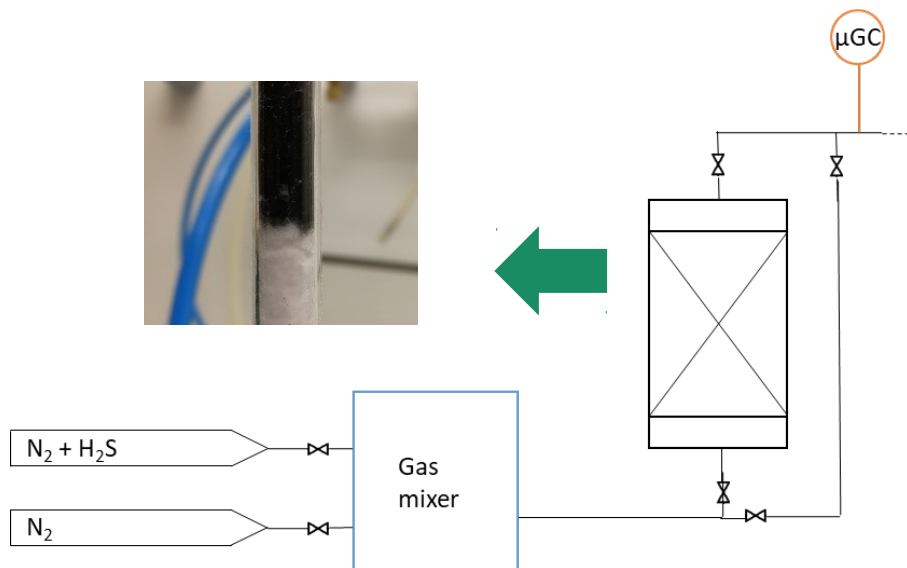



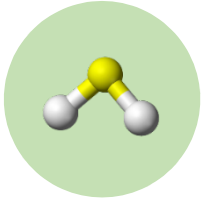
H_2S
Best performing char

2 - Effect of inlet concentration

- Fixed bed reactor - quartz
- Char bed height: 2.5 cm (150 - 200 mg)
- $H_2S + N_2$: 100 NmL/min
- **H_2S : 250 -550 -1000 ppm**
- T_{amb}
- Micro-GC for gas analysis at the

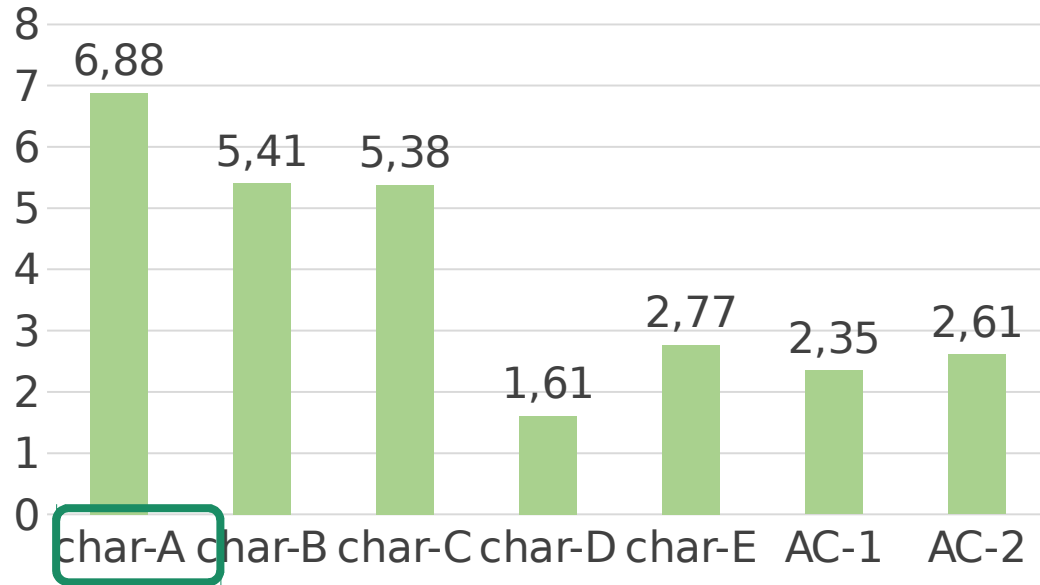
$$m_{ads} = f \frac{M P Q}{R T m_{char}} \int_0^{t_{fin}} (y_{in} - y_{out}) dt$$





Results

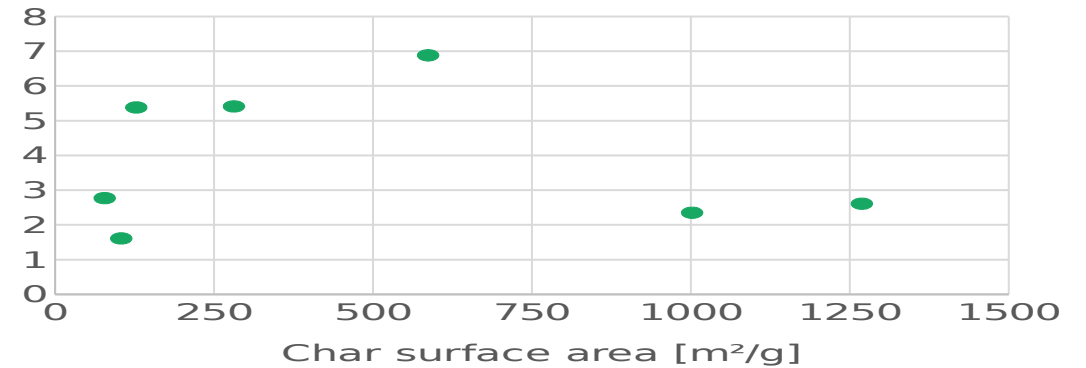
H₂S adsorption capacity [mg/g]



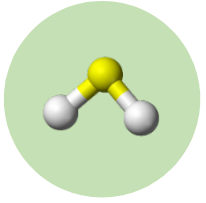
Literature 1.71 - 65 mg/g AC
 0.04 - 0.22 mg/g char from
 pyrolysis

H₂S adsorption capacity [mg/g]

Effect of surface area



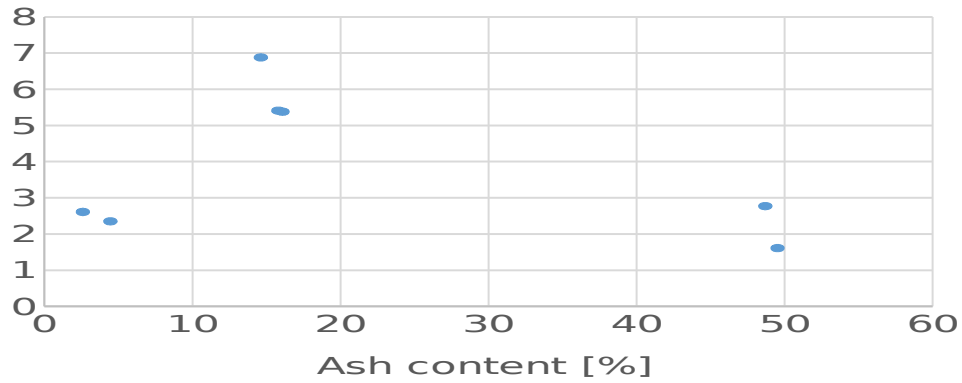
F. Marchelli et al., Experimental study on H₂S adsorption on gasification char under different operative conditions, Biomass Bioenergy 126 (2019) 106 - 116.



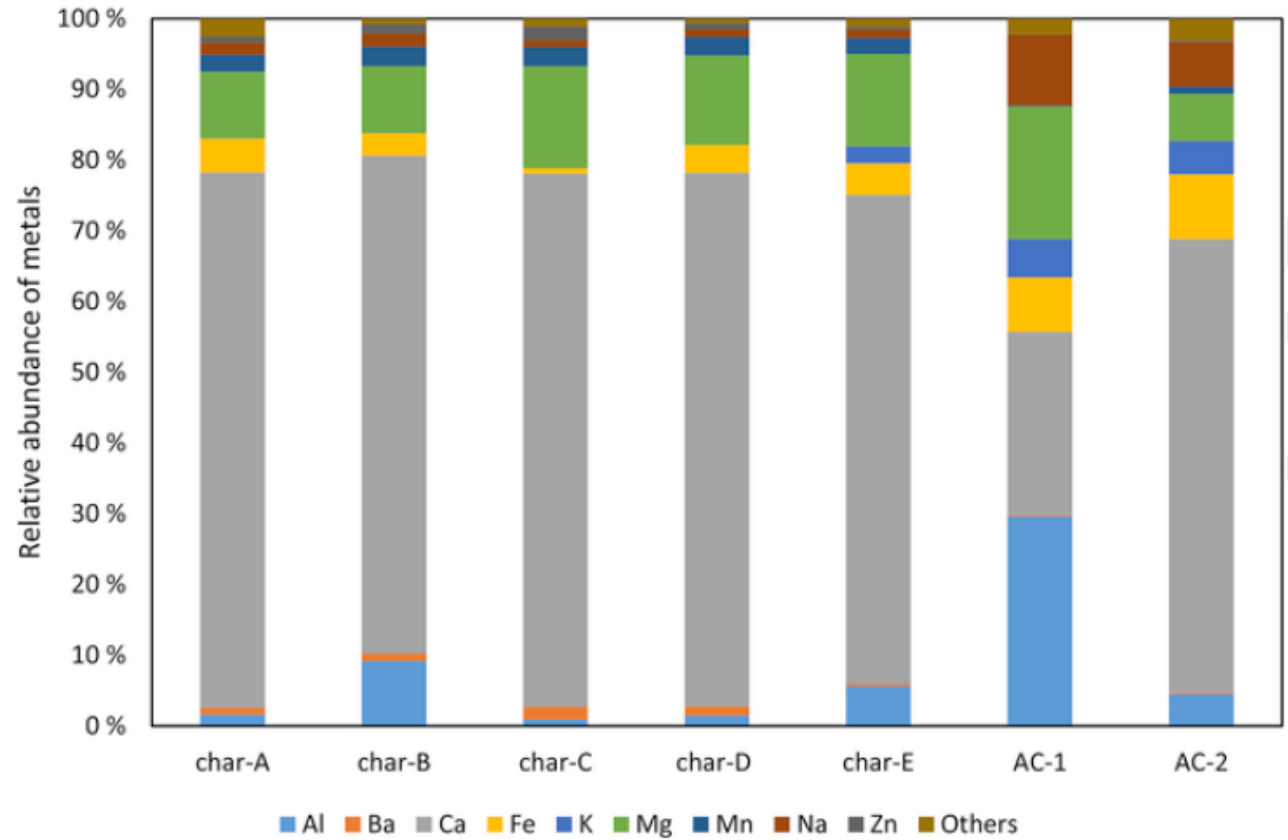
H₂S adsorption capacity [mg/g]

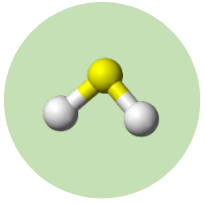
Results

Effect of ash



Metal content

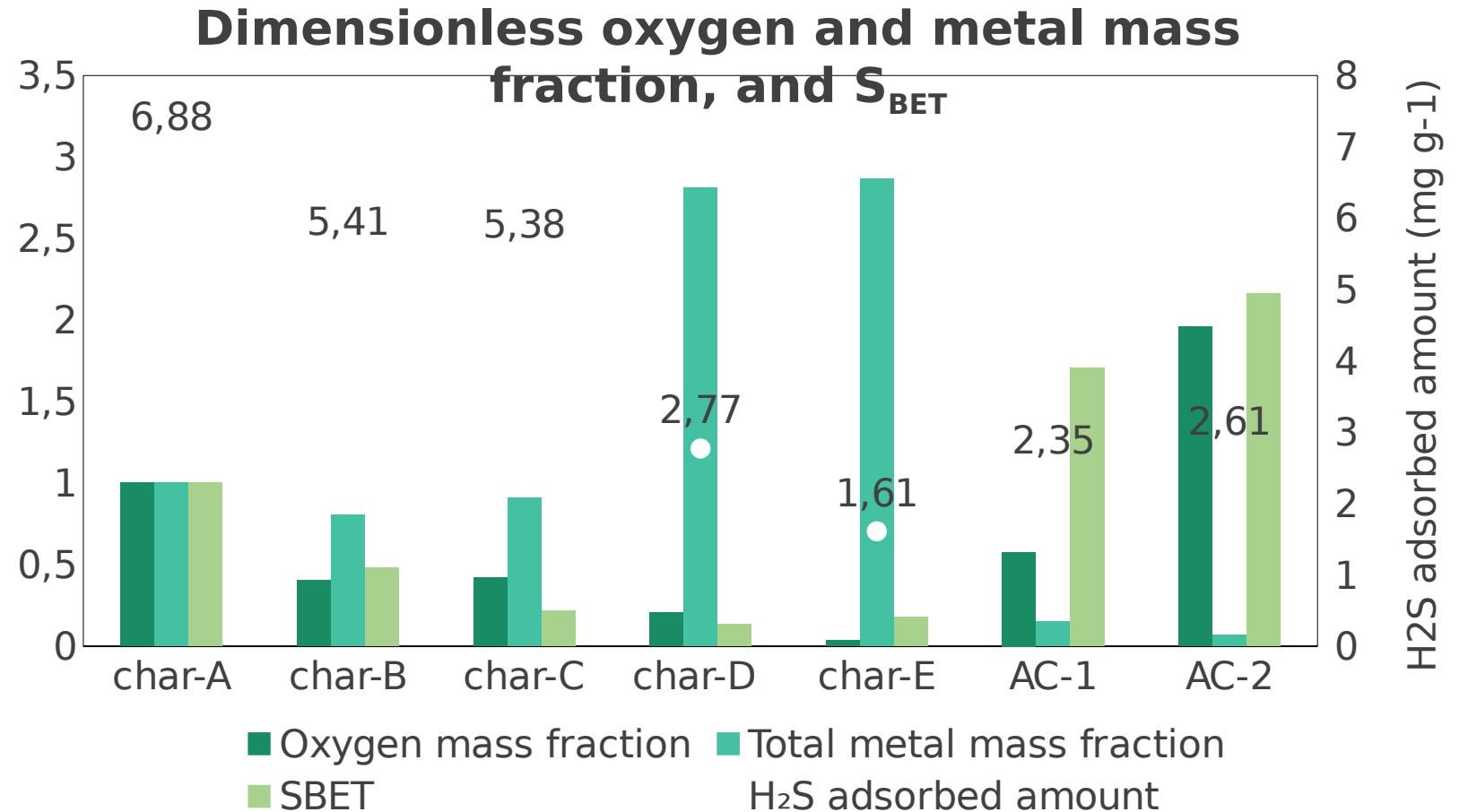
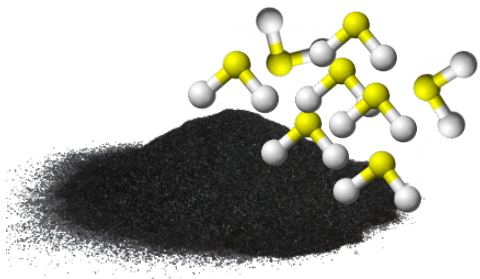


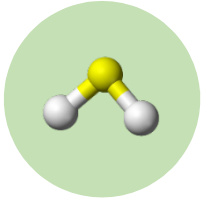


Results

Effects of:

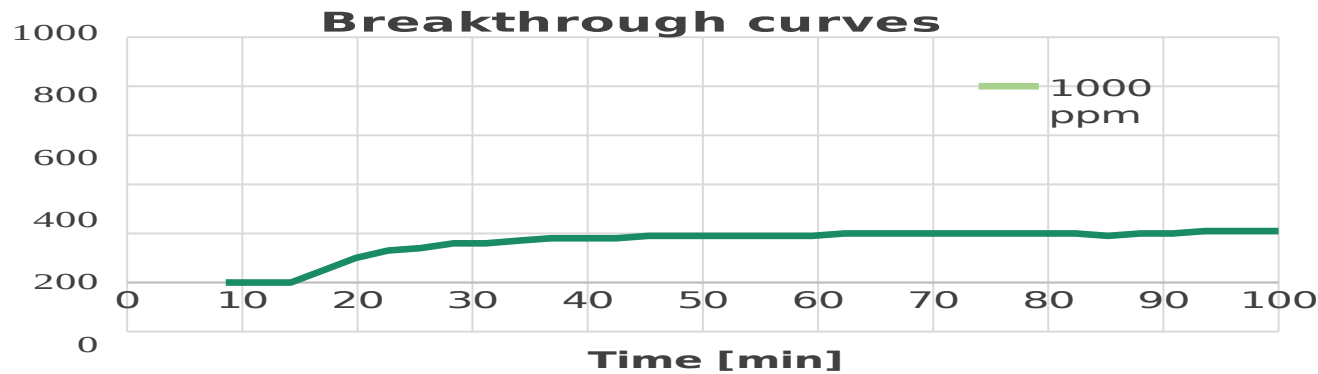
- Oxygen content
- Metals mass fraction
- Surface area





Results - Effect of concentration

H₂S outlet volume fraction (ppm)



Inlet volume fraction of H ₂ S, ppm	H ₂ S adsorption Capacity, mg g ⁻¹
250	6.88 ± 0.37
500	7.87 ± 0.70
1000	6.98 ± 0.24

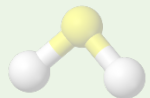
Low concentrations slow down the process, but do not affect the adsorption capacity



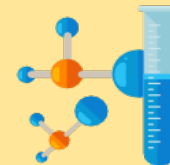
Char collection
and characterization



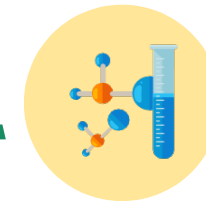
CO₂
adsorption



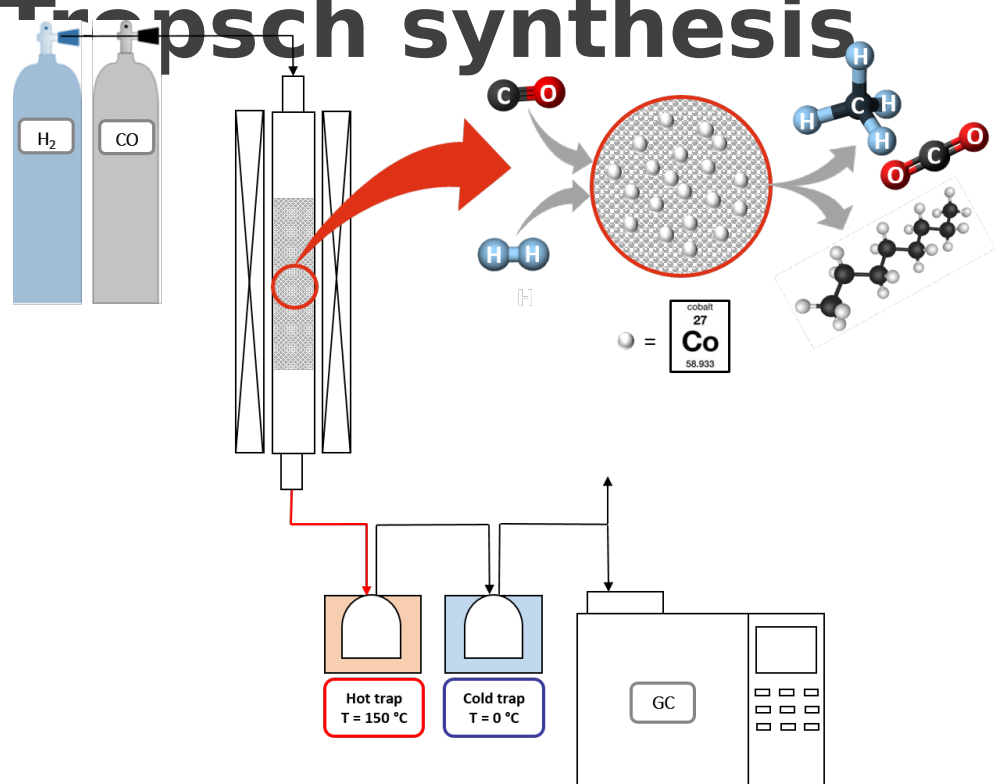
H₂S
adsorption



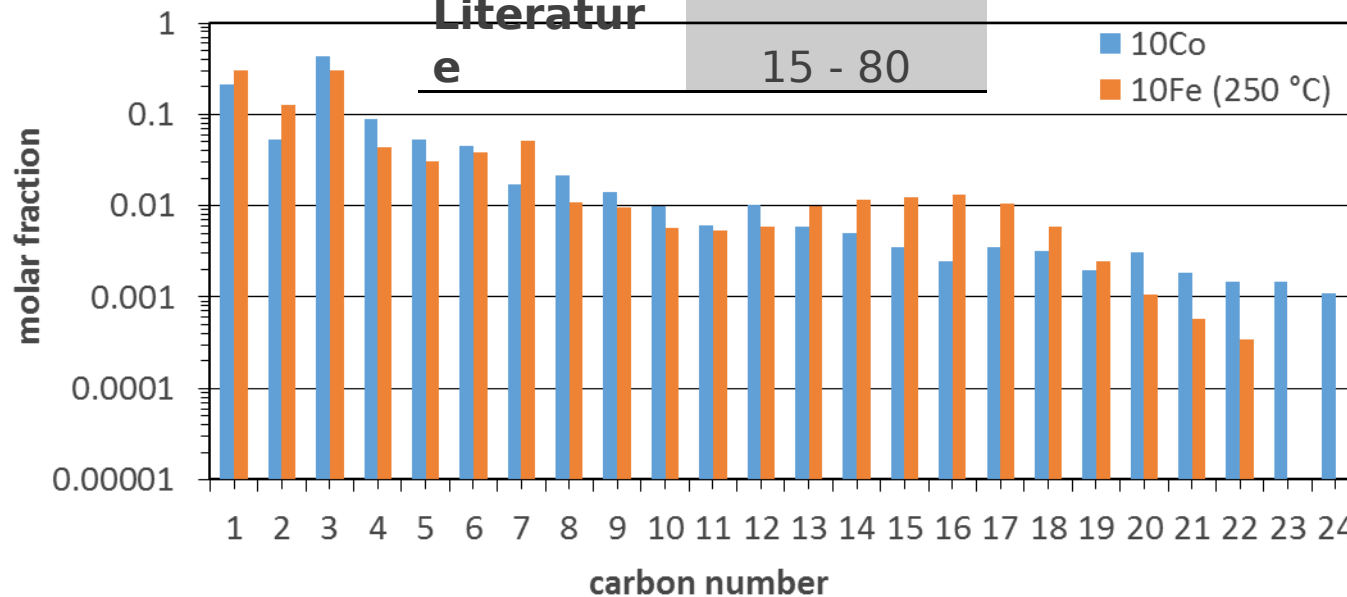
Other
applications



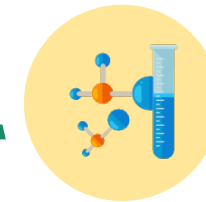
Char as catalyst support for Fischer-Tropsch synthesis



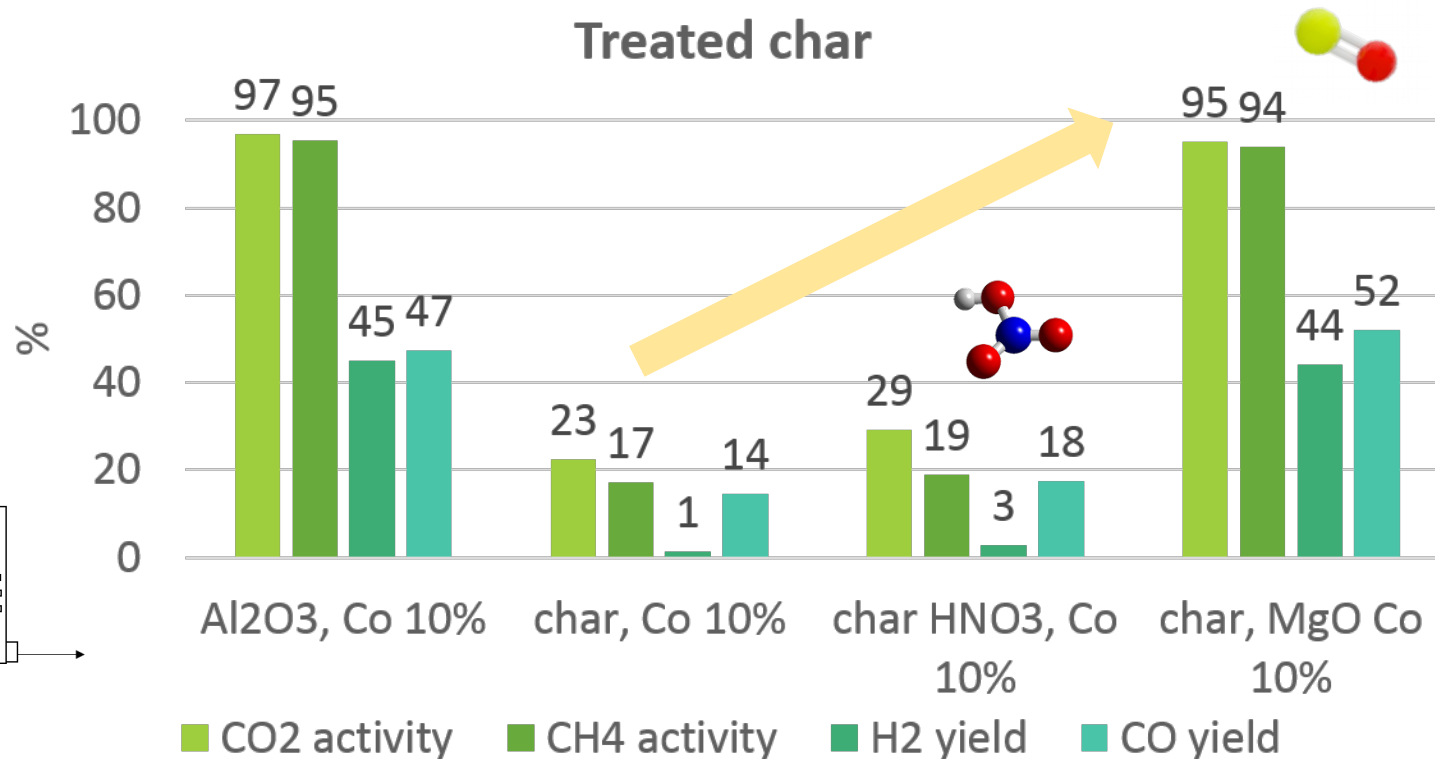
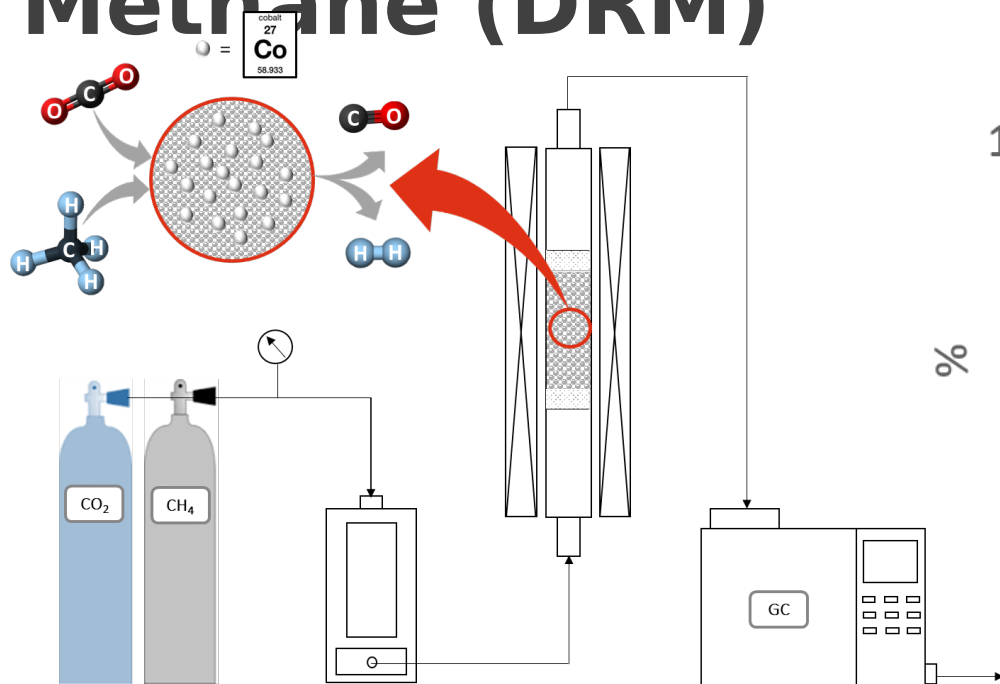
	CO conv., %
Char, 10% Co	8
Char, 10% Fe	26
Literature	15 - 80



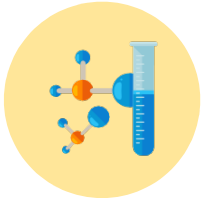
V. Benedetti et al., Investigating the feasibility of valorizing residual char from biomass gasification as catalyst support in Fischer-Tropsch synthesis, under review.



Char as catalyst support for Dry Reforming of Methane (DRM)

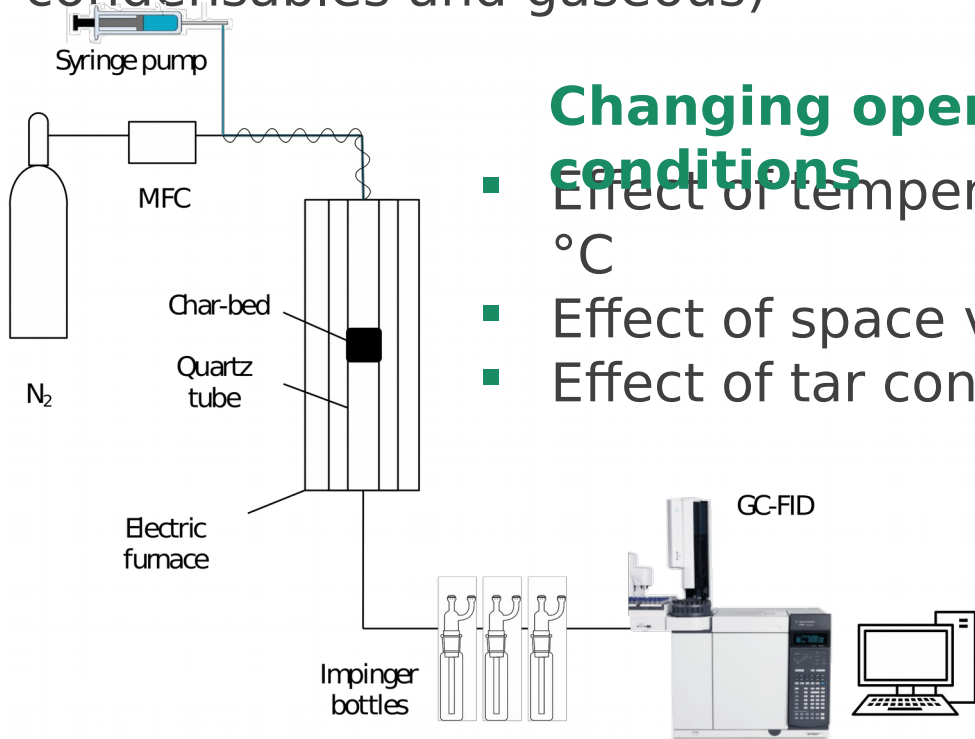


V. Benedetti et al., Valorization of char from biomass gasification as catalyst support in dry reforming of methane, *Front. Chem*, 7 (2018) 119.



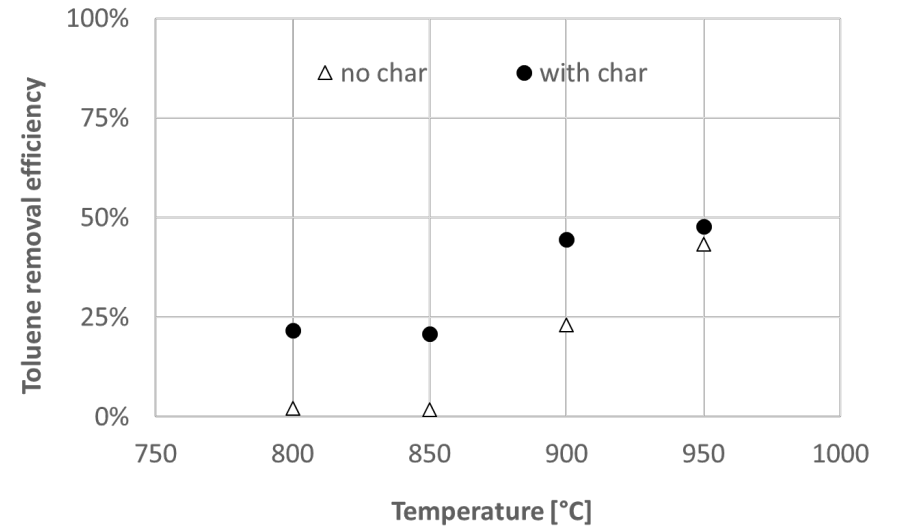
Tar removal by thermal and catalytic cracking

Evaluation of tar removal efficiency and analysis of converted products (both condensables and gaseous)



Changing operating conditions

- Effect of temperature 800-1000 °C
- Effect of space velocity
- Effect of tar concentration



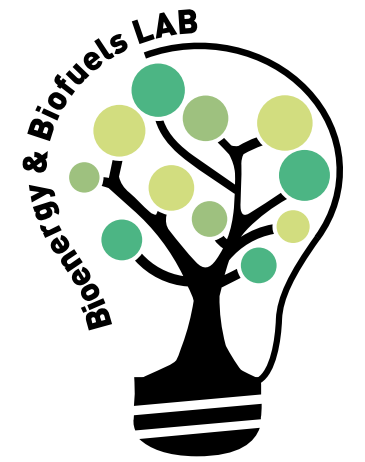
First stage

Model tar compounds:

- Toluene
- Naphthalene
- Phenol

Second stage

Other gas mixtures/real tar from lab-scale gasifier



Thank you for your attention

Valorisation of char residues from biomass gasification in adsorption applications



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vittoria.benedetti@unibz.it

Website:

<https://bnb.groups.unibz.it/>

